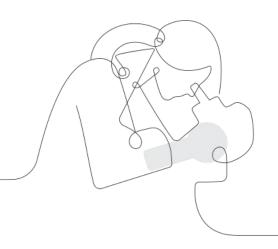
Amplify Science

Grade 7: Chemical Reactions

Applying Reading and Writing Strategies to support Claims, Evidence, and Reasoning within the Amplify Science classroom.

School/District Name
Date
Presented by Your Name



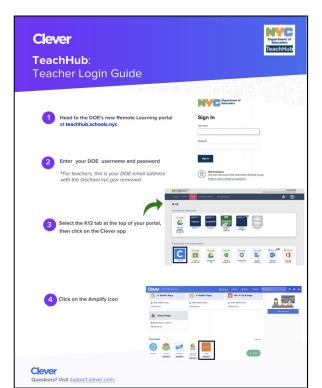
Welcome to Amplify Science! Do Now: Log In

Log In with Amplify

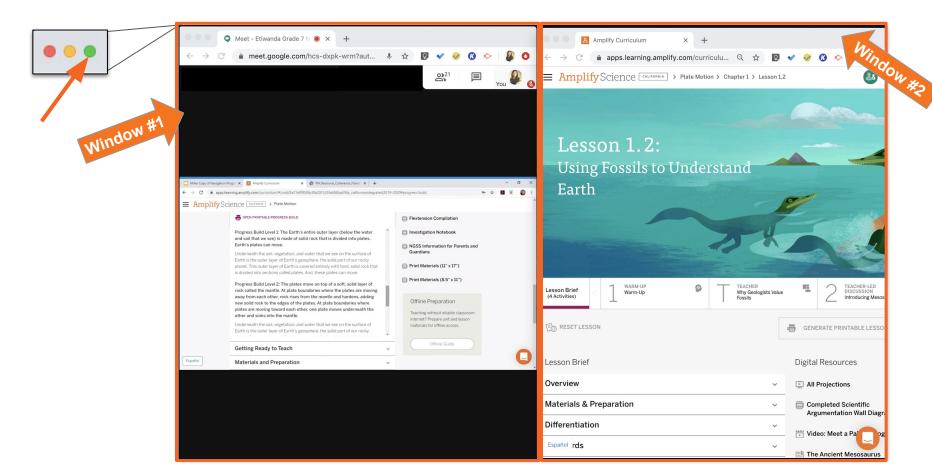




Log in with TeachHUB



Use two windows for today's webinar



Remote Professional Learning Norms



Take some time to orient yourself to the platform

• "Where's the chat box? What are these squares at the top of my screen?. where's the mute button?"



Mute your microphone to reduce background noise unless sharing with the group



The chat box is available for posting questions or responses to during the training



Make sure you have a note-catcher present



Engage at your comfort level - chat, ask questions, discuss, share!

Objectives:

By the end of this workshop, you will be able to:

- Focus on the shifts in emphasizing the Science and Engineering practices which require students to read and write like scientists.
- Identify embedded strategies for explicit literacy support.
- Understand how these strategies help students make and evaluate claims, gather evidence, and justify arguments with relevant reasoning.
- Engage in reflection on how a science seminar structures opportunities for verbal and written argumentation.

Capturing key takeaways!





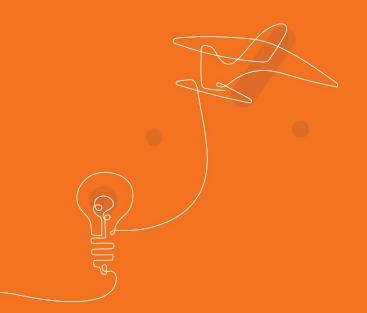
Plan for the day

- Framing the day
 - Welcome and introductions
- Shifts in Science Instruction
 - Revisiting the Amplify Science approach
 - Revisiting Resources
 - Science and Engineering practices
- Science and Literacy
 - Embedded literacy supports
 - Embedded literacy strategies
- Applying the Strategies through the Science Seminar
 - Evaluating claims, gathering evidence, and justifying arguments with reasoning
- Reflection and closing



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Welcome and Introductions

Who's in the Room?

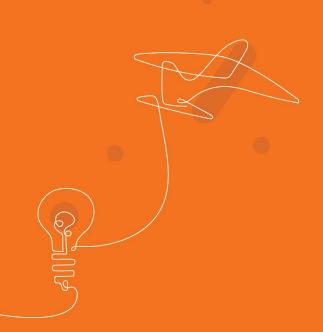
Represent for your borough!



Share your **name, role, & borough**.

Example: Isis, Teacher, 1

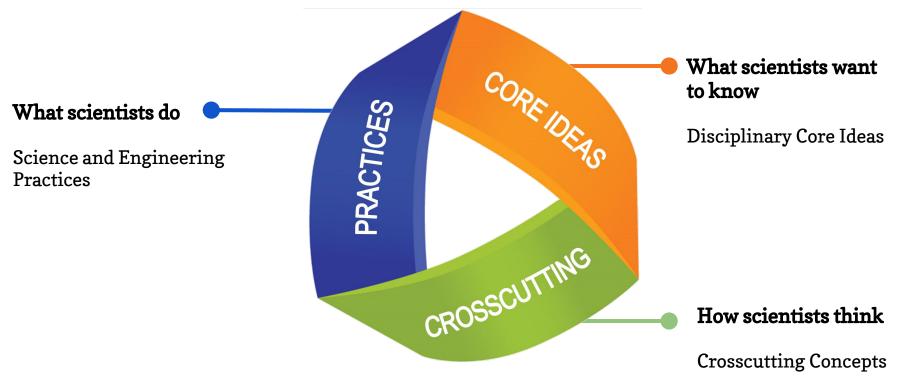
- 1- Brooklyn North
- 2- Brooklyn South
- 3- Queens North
- 4- Queens South
- 5- The Bronx
- 6- Staten Island
- 7- Manhattan



Revisiting the Amplify Science approach

Next Generation Science Standards

Designed to help students build a cohesive understanding of science



Comparing topics and phenomena

A shift in science instruction

from learning about (like a student)



to figuring out

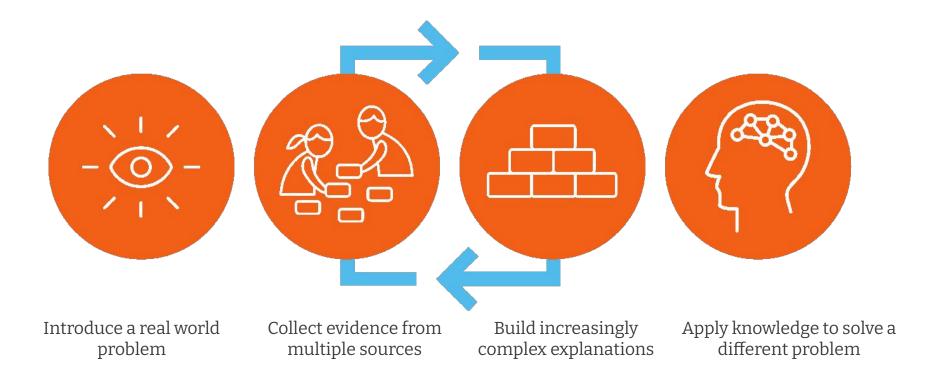
(like a scientist)

Problem-based deep dives

Students inhabit the role of scientists and engineers to explain or predict phenomena. They use what they figure out to solve real-world problems.



Amplify Science approach



Amplify.





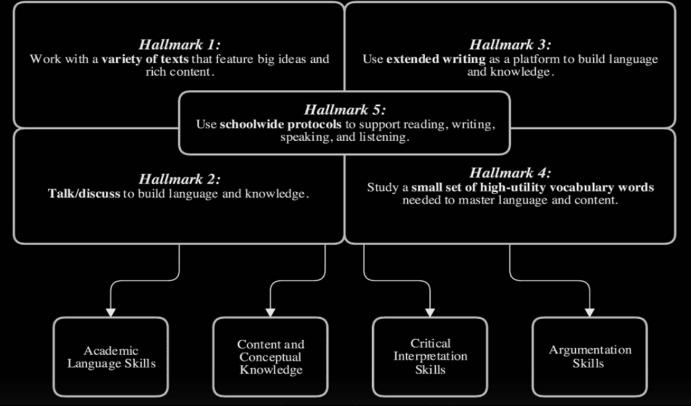


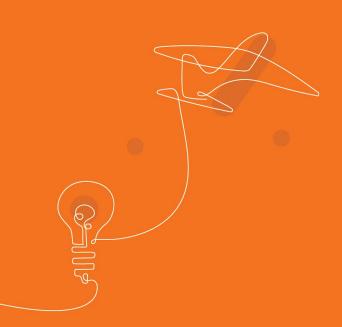




Advanced Literacies

Strengthening the Instructional Core





Revisiting Resources

Middle School Curriculum New York City Edition

Grade 6

- Launch: *
 Harnessing Human
 Energy
- Thermal Energy
- Ocean, Atmosphere, and Climate
- · Weather Patterns
- Populations and Resources
- Matter and Energy in Ecosystems
- Earth's Changing Climate

Grade 7

- Launch: *
 Microbiome
- Metabolism
- · Phase Change
- · Chemical Reactions
- Plate Motion
- Engineering Internship:
 Plate Motion
- Rock Transformations
- Engineering Internship: Earth's Changing Climate

Grade 8

- Launch: Geology on Mars
- Force and Motion
- Engineering Internship:
 Force and Motion
- · Earth, Moon, and Sun
- · Magnetic Fields
- Light Waves
- Traits and Reproduction
- Natural Selection
- · Evolutionary History



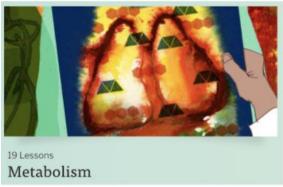
Middle School curriculum: Unit types

Launch Units

Core Units

Engineering Internships







Middle school unit resources



Investigation Notebooks or digital student experience



Teacher's Guide (digital or print)



Articles (digital or print)



Assessments and Reporting



Simulations and other digital tools



Hands-on and print materials

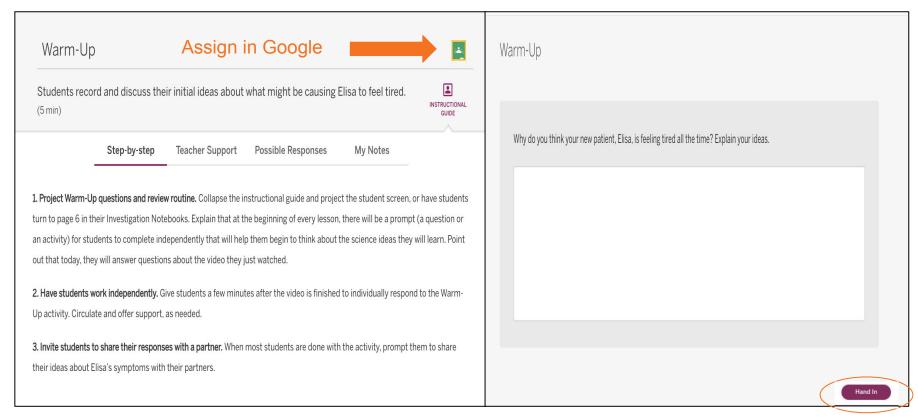


Classroom Slides



Hands-on Flextensions

Middle School Online Component



Welcome Science Educators!

The Amplify Science Program Hub was created to provide you with resources, tools, and advice for all stages of your implementation. Want a tour? Click here!

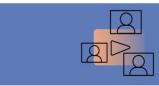
Remote and hybrid learning resources

Amplify Science@Home makes remote and hybrid learning easier.



Professional Learning Resources

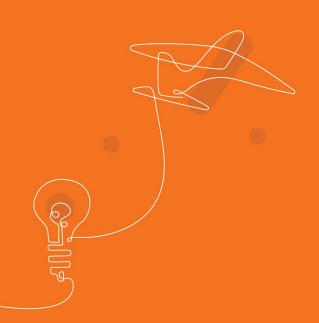
Let's get started!



Additional Unit Materials

Additional resources to complement the units you're teaching.





Science and Engineering Practices



Chemical Reactions

Unit Phenomena:

In the role of student chemists, students explore how new substances are formed as they investigate a problem with the water supply in the fictional town of Westfield. They analyze a reddish-brown substance that is in the water, the iron that the town's pipes are made of, and a substance from fertilizer found to have contaminated the wells that are the source of the town's water, and use their findings to explain the source of the contaminating substance.

Anchor Phenomenon: A mysterious brown substance has been detected in the tap water of Westfield.

Student Role: Students play the role of forensic chemists, applying what they learn about matter and chemical reactions to solve the mystery of mysterious substances appearing in a county's water supply.

Chemical Reactions SEP's

As with all Amplify Science units, the *Chemical Reactions* unit provides students with exposure to most of the science and engineering practices described in the Next Generation Science Standards. This unit particularly emphasizes the following practices:

- **Practice 1: Asking Questions.** As students investigate what caused the problem with the water supply, their inquiry is guided by a series of strategic questions. They also have many opportunities to pose their own questions. In particular, the Active Reading approach, an approach to reading based on curiosity and inquiry, supports students in asking thoughtful questions as they read science articles.
- **Practice 2: Developing and Using Models.** Students spend ample time exploring and investigating a digital simulation that models how substances behave at the atomic scale during chemical reactions. They also create visual models that show their ideas about how substances form during chemical reactions.
- Practice 3: Planning and Carrying Out Investigations. Students conduct investigations using the digital simulation and with real substances to develop an understanding of properties of substances and of chemical reactions.
- **Practice 4: Analyzing and Interpreting Data.** Students analyze data about the properties of substances and data about the atoms that make up substances.
- **Practice 6: Constructing Explanations.** Students spend time discussing, writing, and creating models to aid them in developing their explanations of concepts presented in the unit.
- Practice 7: Engaging in Argument from Evidence. Students receive instruction about the structure of a scientific argument and are supported in evaluating evidence, engaging in scientific reasoning, and producing written arguments.
- Practice 8: Obtaining, Evaluating, and Communicating Information. Students have multiple opportunities to
 engage in Active Reading, an approach to obtaining information from science texts. Students also evaluate
 evidence to determine its strength.

Chemical Reactions

Multiples Modalities

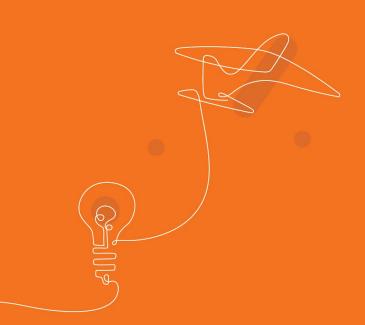
- **Do.** Students investigate and observe substances at the atomic scale, using the *Chemical Reactions* Simulation. On several occasions, they use this information to explain an observable change in appearance they witnessed during a hands-on activity.
- **Talk.** There are multiple opportunities to engage in student-to-student and whole-class discussions that reflect on the connection between the changes that can be observed and what is happening at the atomic scale during a chemical reaction.
- **Read.** In the "Atomic Zoom-In" and "What Happens When Fuels Burn?" articles, students read about the connection between the atomic scale and how it relates to what can be observed.
- Write. Students write scientific arguments explaining how changes at the atomic scale lead to the observable appearance of rust and another substance in Westfield's water supply.
- **Visualize.** Students represent their ideas about chemical reactions by creating visual atomic- scale models in the *Chemical Reactions* Modeling Tool that help explain observable phenomena.



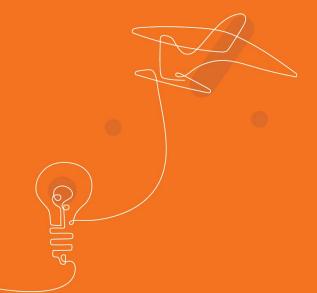


Plan for the day

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Science and Literacy



Embedded Literacy Supports

AmplifyScience

Amplify Science

Welcome

Program developers

Designed for the NGSS

Program components

Scope and Sequence

Phenomena, standards, and progressions

Assessments

Science and literacy

Access and equity

Resources

Welcome

The Program Guide details information about the program, including its authorship, development, themes, and more. It serves as a resource for finding out more about the program's structure, components, supports, how it meets standards, and flexibility.

Navigate through the links on the left-hand side of the page to access more information about the program and to explore resources that can help with your implementation.

ACCESS THE DIGITAL CURRICULUM

Support

....

Amplify Help Center

scihelp@amplify.com

1-800-823-1969

More Amplify Science

Transitional Kindergarten (TK)

Search Site ...

Amplify.

Terms of service Privacy **Learn about Amplify**

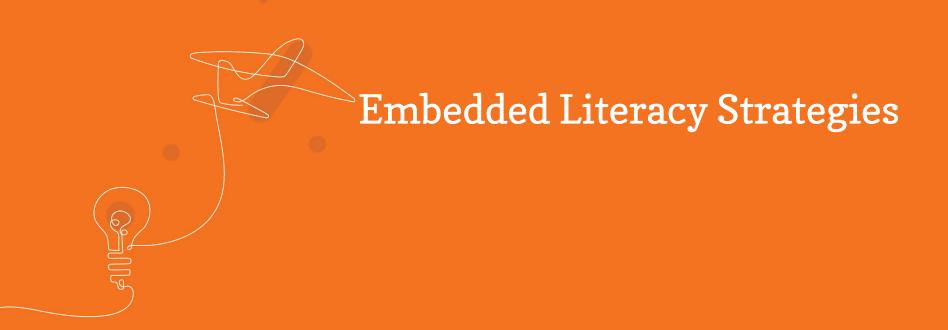
Company / Leadership / Partners / Newsroom

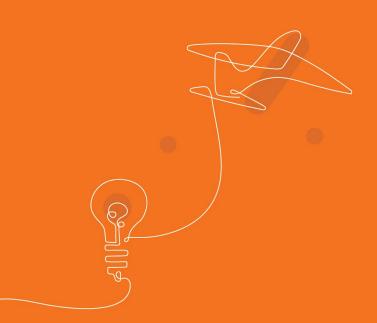
Connect with Amplify

Contact / Support / Events / Careers

Keep up with Amplify

Twitter / Facebook / LinkedIn / YouTube / Google





Active Reading



Reading in middle school science

Amplify.

Reflect and Share

- What has been a challenge of reading in Amplify Science in your classroom? What has been a success?
- What strategies do you use to engage and support your students with reading in Amplify Science while teaching remotely?

Active Reading

 Active reading is explicitly introduced in the Launch Unit for each grade level.

Available formats:

- Article Compilation (PDF)
- Digital Articles (digital library)
- Student Editions (hardcover books)





Active Reading Instructional Approach

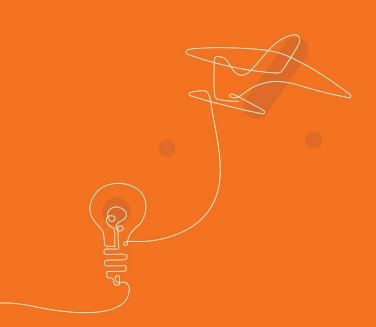
A typical Active Reading sequence

First Read Second Read Third Read

Independent, followed by paired and whole class discussion

Reading for a teacher-directed purpose, followed by a paired, complementary activity

Diving into the text for other, content-related purposes



Active Reading Components

A typical Active Reading sequence

First Read Second Read Third Read

Independent, followed by paired and whole class discussion

Reading for a teacher-directed purpose, followed by a paired, complementary activity

Diving into the text for other, content-related purposes

Text Roles/Functions for Inquiry-Based Investigations

Scientists read and interpret others' data and findings

Illustrate phenomenon that would otherwise be

Provide context for inquiry-based

investigations

Deliver content

	unobservable; opportunities to apply what students are learning
Model scientific processes	Model inquiry processes; Modeling scientific dispositions; Depicting scientists and their work
Support secondhand investigations (collection of textual data)	Provides data for interpretation represented with graphs, pictures, tables; communicating visuals information based in data
Support first-hand investigations (collection of empirical data)	Providing students information to supplement their empirical (first-hand) studies; Support the design and implementation of investigations.
From Cervetti, G. N. & Barber, J. (2009). Text in ha	ands-on science. In Hiebert, E. H. & Sailors, M. (Eds.) Finding the Right Texts: What Works

for Beginning and Struggling Readers. New York: The Guilford Press.



Active Reading Guidelines

- Think carefully about what you read. Pay attention to your own understanding.
- 2. As you read, annotate the text to make a record of your thinking. Highlight challenging words and add notes to record questions and make connections to your own experience.
- **3.** Examine all visual representations carefully. Consider how they go together with the text.
- **4.** After you read, discuss what you have read with others to help you better understand the text.

Exploring an Active Reading Sequence

Directions:

- Navigate to the Chemical Reactions unit
- Scroll down to the Unit Guide
- Click "Articles in This Unit"
- Choose an article
- Fill out the sheet on page 6 of your participant notebook

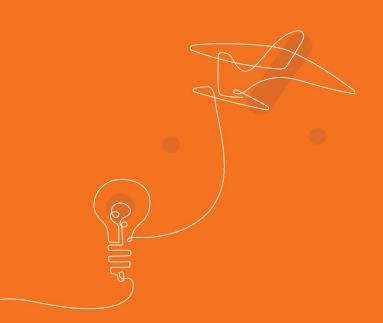
Exploring an Active Reading Sequence

Directions:

- · Navigate to your current unit
- · Scroll down to the Unit Guide
- · Click Articles in This Unit
- Choose an article
- Fill out this sheet

What is the article about?		Article Title:	
	First Read		
What is the purpose of this read?	What are students doing as they read? How are they supported?	How does this build on students' unit-level understanding?	
	Second Read		
What is the purpose of this read?	What are students doing as they read? How are they supported?	How does this build on students' unit-level understanding?	
	Third Read		
What is the purpose of this read?	What are students doing as they read? How are they supported?	How does this build on students' unit-level understanding?	

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Writing

Reflect and Share

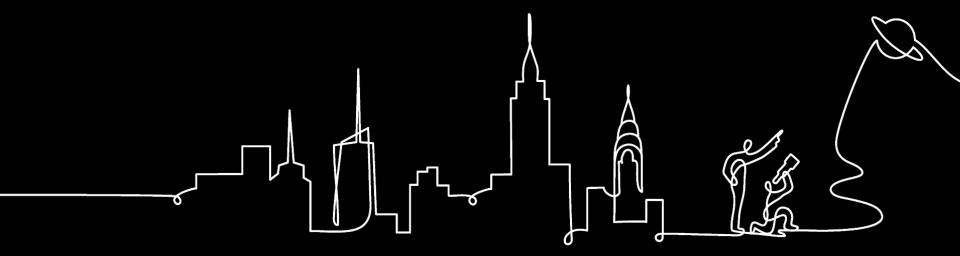
What is the purpose of writing in Amplify Science?

What has been a challenge of writing in Amplify Science?

 What strategies do you use to engage and support your students with writing in Amplify Science in a remote/hybrid setting?

Writing in Amplify Science

Purposeful communicative writing is an integral part of the Amplify Science curriculum. Students write daily for many different purposes.



Completed Scientific Argumentation Wall Diagram

Scientific Argumentation

* These items were posted in the launch unit.

The purpose of a scientific argument is to convince others, using evidence and reasoning.

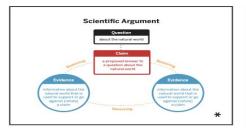
Evaluating Evidence

*

Evidence criteria posted in previous units

Evidence is higher quality if it comes from a reliable source.

3.2



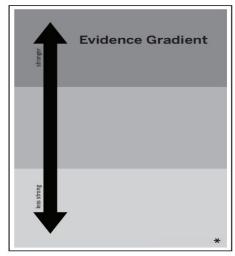
A scientific argument . . .

- begins with a question.
- has a claim that proposes an answer to the question.
- · has evidence that supports the claim.
- clearly explains how the evidence supports the claim (reasoning).

Evidence	This matters because (How does this evidence support the claim?)	Therefore, (claim)
		-
		,

Argumentation Sentence Starters

- I think this evidence supports this claim because . . .
- I don't think this evidence supports this claim because . . .
- I agree because . . .
- · I disagree because . . .
- · Why do you think that?



Example Student Arguments

Reasoning Tool

Evidence	This matters because (How does this evidence support the claim?)	Therefore, (claim)



Using the Reasoning Tool to Support Your Claim

- Circle your strongest piece of evidence.
- Draw an X over those pieces of evidence that you do not plan to use in your argument.
- Draw an arrow to connect pieces of evidence that go together.

Example		
Evidence	This matters because (How does this evidence support the claim?)	Therefore, (claim)
Sample Evidence Card A	Your ideas about how the evidence supports the claim	Your claim
Sample Evidence Card B	Your ideas about now the evidence supports the claim	
Sample Evidence Card C	Your ideas about how the evidence supports the claim	

Exploring Writing Sequence

Directions:

- Download your units investigation notebook from Printable Resources in the Unit Guide for chapter 1. Use this to help you identify opportunities for students to write
- Analyze the purpose of each writing opportunity in the chapter.
- Fill out the sheet on page 7 of your participant notebook

Unit: _		Chapter:
Gu	ownload your unit's Investigation Notebook uide. Use this to help you identify opportur halyze the purpose of each writing opportu a. Record the activity and lesson in the fi b. If the purpose of activity is unclear fro	ities for students to write. nity in a chapter. irst column. m the Investigation Notebook page, use n Overview Compilation, or navigate to
ctivity	Purpose for the student to write	Purpose for the teacher • How will you find this helpful?

Amplify.

5 min break





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Science Seminar Sequence







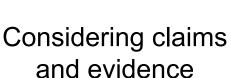
Considering claims and evidence

Participating in the Science Seminar

Writing an argument

Science Seminar Sequence







Participating in the Science Seminar



Writing an argument

Goals for the Science Seminar Sequence

- Apply content knowledge (DCI's and CCC's) gained throughout the unit to address a new scientific problem
- Highlight practices: making arguments from evidence, constructing explanations, analyzing data, communicating information
- Three-dimensional assessment opportunity
- Engagement: student-centered, open-ended, novel context
- Nature of science: questions with no clear answer



Claims Students Investigate

Claim 1: Pat is most likely to have made the hydrofluoric acid by using sulfuric acid and calcium fluoride.

Claim 2: Alex is most likely to have made the hydrofluoric acid by using sulfuric acid and magnesium chloride.

Claim 3: Tracy is most likely to have made the hydrofluoric acid by using purified water and fluorine.

To: Student Chemists

From: Dr. Samara Yung, Lead Chemist

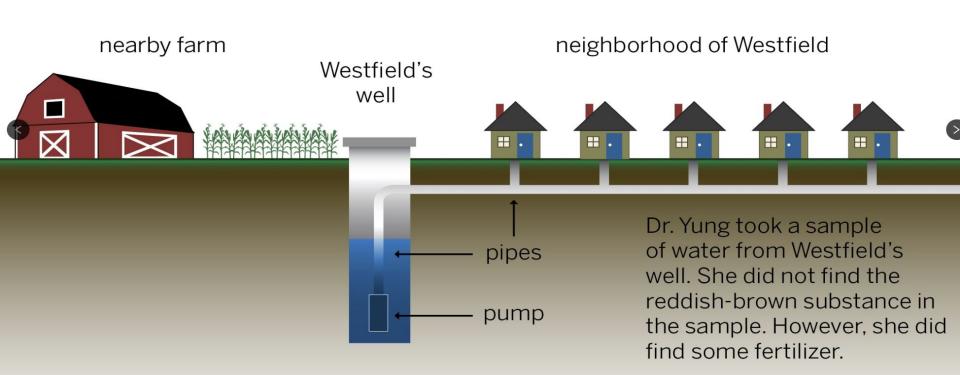
Subject: Someone Stole the Lavoisier Diamond!



There's been a robbery at the Westfield Museum! This morning, the museum staff discovered the rare and expensive Lavoisier Diamond was missing. When police arrived, they found a hole had been made in the glass display case where the diamond was stored. Next to the display case, they found a plastic container that was half full with an unknown substance. Police think that the thief may have used some of the substance to make a hole in the glass and may have accidentally left the rest of the substance behind.

The chief of police has asked me to help them identify this unknown substance and determine who could have used it to steal the Lavoisier Diamond. As you have recent experience in working with the community of Westfield, I am turning the case over to you. I hope you remember what you have learned so far about chemical reactions. I think you will need to use your knowledge to help the police crack this case!

The residents of Westfield found a reddish-brown substance coming out of their water pipes.



Evaluating Observations of the Unknown Substance

There were four police officers at the crime scene. The officers wrote down their observations of the unknown substance that was used to make a hole in the glass.

- 1. With a partner, read the observations written down on the Unknown Substance Evidence Cards given to you by your teacher.

 Annotate the cards with any questions or ideas you have.
- 2. Discuss the cards with your partner and evaluate each observation using the Evidence Criterion included below.
- 3. Once you have evaluated each observation, place the cards on the Evidence Gradient sheet with the strongest pieces of evidence near the top and the less strong pieces of evidence near the bottom.
- 4. When you are finished, answer the questions below.

Evidence Criterion: More detailed observations provide stronger evidence.

Evaluating Evidence

Officer Hodges's observation:

"To me, it looked like water but a little different."

Chemical Reactions—Unknown Substance Evidence Cards—Lesson 4.1 © 2016 The Regents of the University of California

Officer Lee's observation:

"At room temperature, it was a liquid, and it smelled awful."

Chemical Reactions—Unknown Substance Evidence Cards—Lesson 4.1 © 2016 The Regents of the University of California

Officer Diaz's observation:

"It was a colorless liquid with a strong, irritating odor."

Chemical Reactions—Unknown Substance Evidence Cards—Lesson 4.1

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Officer Williams's observation:

"I saw a clear liquid that seemed to be dangerous."

Chemical Reactions—Unknown Substance Evidence Cards—Lesson 4.1

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Officer Hodges's observation:

"To me, it looked like water but a little different."

Chemical Reactions—Unknown Substance Evidence Cards—Lesson 4.1

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Officer Lee's observation:

"At room temperature, it was a liquid, and it smelled awful."

Chemical Reactions—Unknown Substance Evidence Cards—Lesson 4.1

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Officer Diaz's observation:

"It was a colorless liquid with a strong, irritating odor."

Chemical Reactions—Unknown Substance Evidence Cards—Lesson 4.1

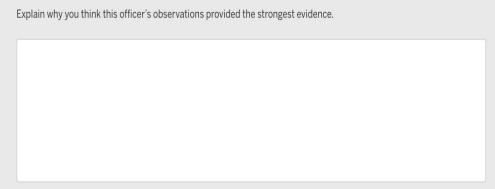
© 2016 The Regents of the University of California

Officer Williams's observation:

"I saw a clear liquid that seemed to be dangerous."

Chemical Reactions—Unknown Substance Evidence Cards—Lesson 4.1
© 2016 The Regents of the University of California

Which officer provided the strongest evidence?
Officer Hodges
Officer Lee
Officer Diaz
Officer Williams



List of Corrosive Substances

Substance name	Properties
hydrobromic acid	faint yellow colorstrong, irritating odorliquid at room temperature
hydrofluoric acid	colorlessstrong, irritating odorliquid at room temperature
iodine monochloride	 dark red color strong, irritating odor liquid at room temperature
perchloric acid	colorlessodorlessliquid at room temperature

Which of these substances do you think is the unknown substance the criminal used to make a hole in the glass?
Hydrobromic acid
Hydrofluoric acid
lodine monochloride
Perchloric acid
Explain your answer.

Making Hydrofluoric Acid

Goal: Create atomic-scale models that show whether or not each of the suspects could have produced hydrofluoric acid. H(F Key Before After Pat hydrogen calcium fluoride sulfuric acid oxygen **Before** After Alex chlorine sulfuric acid magnesium chloride magnesium **Before** After fluorine **Tracy** calcium purified water

Sorting Evidence

Evidence Pat made the hydrofluoric acid:	Evidence Tracy made the hydrofluoric acid:	Evidence Card A	Evidence Card B	Evidence Card C	Evidence Card D
		When the police searched Pat's house, they found some calcium sulfate (CaSO ₄).	The police did not find hydrofluoric acid (HF), sulfuric acid (H ₂ SO ₂), or calcium fluoride (CaF ₂) in Pat's house.	When the police searched Tracy's house, they found some purified water (H ₂ 0).	The police did not find hydrofluoric acid (HF), fluorine (F ₂), or any other unusual substances in Tracy's house.
		Ca S 0	H F H O O S O H	HOH	HF FF
		Chemical Reactions—Suspect Evidence Cards—Lesson 4.2 © 2016 The Regents of the University of California	Chemical Reactions—Suspect Evidence Cards—Lesson 4.2 © 2016 The Regents of the University of California	Chemical Reactions—Suspect Evidence Cards—Lesson 4.2 © 2016 The Regents of the University of California	
Evidence Pat did not make the hydrofluoric acid:	Evidence Tracy did not make the hydrofluoric acid:	Evidence Card E	Evidence Card F	Evidence Card G	Evidence Card H
		According to Dr. Yung, calcium sulfate (CaSO ₄) is a substance commonly used in sculpting.	According to Dr. Yung, oxygen (O_2) is a colorless, odorless gas that is commonly found in air.	The chemical supply company confirmed the delivery of sulfuric acid (H ₂ SO ₄) and calcium fluoride (CaF ₂) to Pat's house.	The chemical supply company confirmed the delivery of purified water (H ₂ O) and fluorine (F ₂) to Tracy's house.
		(Ca) (S) (O)	00	F Ca F O O H	HOH FF
	Chemical Reactions—Evidence Sorting Grid—Lesson 4.2 © 2016 The Regents of the University of California	Chemical Reactions—Suspect Evidence Cards—Lesson 4.2 © 2016 The Regents of the University of California	Chemical Reactions—Suspect Evidence Cards—Lesson 4.2 © 2016 The Regents of the University of California	Chemical Reactions—Suspect Evidence Cards—Lesson 4.2 © 2016 The Regents of the University of California	Chemical Reactions—Suspect Evidence Cards—Lesson 4.2 © 2015 The Regents of the University of California

Sorting New Evidence

Sorting Through New Evidence

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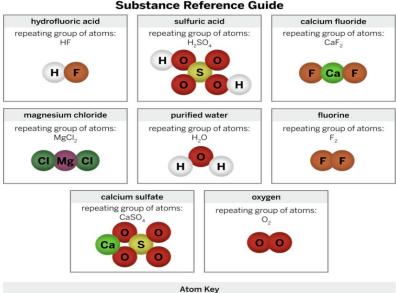
hydrogen

Based on your models, the police obtained search warrants for Pat and Tracy's houses. They were able to find new evidence about the substances in each of their houses.

Instructions:

- 1. Examine the Suspect Evidence Cards, annotating with any questions or ideas you have. If useful, you may also color in the atoms on the cards using the atom key below.
- 2. Review the Modeling Tool sheet you completed in the last activity. Consider how the information in the Suspect Evidence Cards might relate to your completed Modeling Tool sheet. You may revise your model if needed.
- 3. As you sort the cards using the Evidence Sorting Grid sheet given to you by your teacher, discuss your choices with your partner. If you think a card belongs in more than one category, place it on the line between those two categories. If you are not sure which category a card belongs in, set that card to the side

Substance Reference Guide



chlorine

magnesium

fluorine

sulfur

Amplify.

Ca

calcium

Identifying the Primary Suspect

Based on how you sorted the Suspect Evidence Cards with your partner, answer the questions below. Remember that you will have a chance to change your mind in the next lesson.

Question: Which suspect is most likely to have made the hydrofluoric acid?

Claim 1: Pat is most likely to have made the hydrofluoric acid by using sulfuric acid and calcium fluoride.



Claim 2: Alex is most likely to have made the hydrofluoric acid by using sulfuric acid and magnesium chloride.



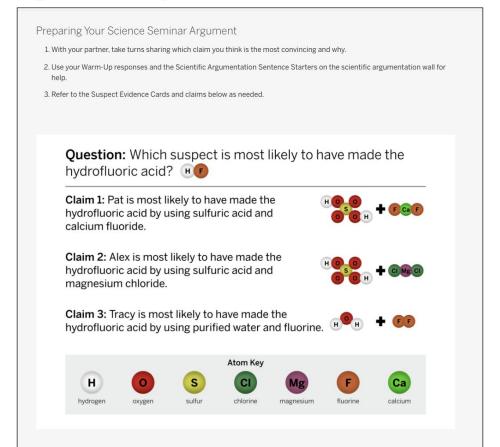
Claim 3: Tracy is most likely to have made the hydrofluoric acid by using purified water and fluorine.





ydrofluoric aci	ou know so far, which suspect do you think is most likely to have made the 1?
Pat	
Alex	
Tracy	
Not sure	
xplain your answe	r using at least one piece of evidence from your Suspect Evidence Cards.

Preparing for the Seminar



Science Seminar Sequence



Considering claims and evidence



Participating in the Science Seminar



Writing an argument

Science Seminar expectations

Students are expected to:

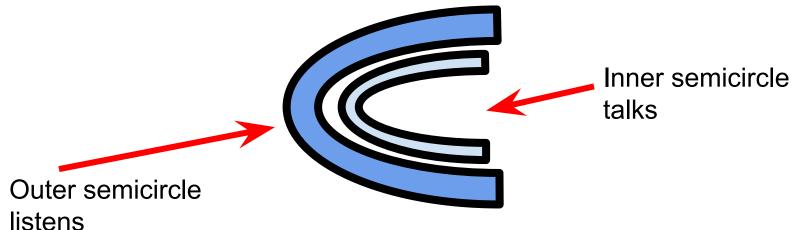
- Run the conversation.
- Use evidence to support ideas.
- Explain their thinking.
- Listen to one another.
- Respond to one another.
- Be open to changing their minds.



Science Seminar seating

Class arrangement:

- Half the class sits in the inner semicircle.
- The other half of the class sits in the outer semicircle.



Science Seminar Observations

Write a check mark in the right-hand column every time you hear one of your peers say or do something listed in the left-hand column. If you hear an interesting idea, write it in the last row of the table.

OBSERVATIONS DURING THE SEMINAR CHECK MARKS I heard a student use evidence to support a claim. I heard a student respectfully disagree with someone else's thinking. I heard a student explain how her evidence is connected to her claim. I heard a student evaluate the quality of evidence. I heard an idea that makes me better understand one of the claims. That idea is:

Scaffolding Talk

Add a new idea with evidence:

- I think _____ because...
- My evidence is...

Agree/Disagree and use evidence:

- I agree/disagree with _____
 because...
- I am now convinced that ____ because...

Ask a question

- What is your evidence?
- Given this evidence, how sure are you? How could you be more sure?
- Do you agree or disagree with what _____ said?
- I have a question for _____about...
- Could you say more about that?
- Could you give us an example?
- I wonder...

Amplify.

Participating in the Seminar



Evidence Card A

When the police searched Pat's house, they found some calcium sulfate (CaSO₄).



Evidence Card B

The police did **not** find hydrofluoric acid (HF), sulfuric acid (H₂SO₄), or calcium fluoride (CaF₂) in Pat's house.



Evidence Card C

When the police searched Tracy's house, they found some purified water (H₂O).



Evidence Card D

The police did **not** find hydrofluoric acid (HF), fluorine (F_2) , or any other unusual substances in Tracy's house.



Evidence Card E

According to Dr. Yung, calcium sulfate ($CaSO_4$) is a substance commonly used in sculpting.



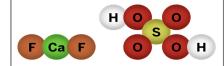
Evidence Card F

According to Dr. Yung, oxygen (O_2) is a colorless, odorless gas that is commonly found in air.



Evidence Card G

The chemical supply company confirmed the delivery of sulfuric acid (H₂SO₄) and calcium fluoride (CaF₂) to Pat's house.



Evidence Card H

The chemical supply company confirmed the delivery of purified water (H_2O) and fluorine (F_2) to Tracy's house.





Science Seminar Sequence







Participating in the Science Seminar



Writing an argument

Using the Reasoning Tool to Support Your Claim

- Circle your strongest piece of evidence.
- Draw an X over those pieces of evidence that you do not plan to use in your argument.
- Draw an arrow to connect pieces of evidence that go together.

Example		
Evidence	This matters because (How does this evidence support the claim?)	Therefore, (claim)
Sample Evidence Card A	Your ideas about how the evidence supports the claim	Your claim
Sample Evidence Card B	Your ideas about now the evidence supports the claim	
Sample Evidence Card C	Your ideas about how the evidence supports the claim	

Writing a Scientific Argument

willing a S	Scientific	Argumer	IT Name			Date
Get ready to writ a scientific argur					swered the qu	uestions, wri
Question: Whi	ich suspect is r	most likely to	have made t	ne hydrofluorid	acid?	F
Claim 1: Pat is using sulfuric			e hydrofluor	ic acid by (H)	9 9 9 0 0 4	FCaF
Claim 2: Alex by using sulfu				ric acid H	S O H	CI Mg CI
					_	
Claim 3: Tracy by using purifi				oric acid (HOH +	FF
by using purifi	ed water and f	luorine.	Atom Key			
				oric acid (Mg)	F fluorine	Ca calcium
by using purifi	ed water and fl	S sulfur	Atom Key CI chlorine	Mg magnesium	F	Ca
by using purifi	o oxygen are you going to the control of the contr	luorine. S sulfur o make in you	Atom Key CI chlorine	Mg magnesium	F fluorine	Ca
by using purification by using purification by the bound of the bound	o oxygen are you going to the control of the contr	luorine. S sulfur o make in you ly to have mad	Atom Key CI chlorine ar argument?	Mg magnesium	F fluorine	Ca calcium

	Review the Suspect Evidence Cards and the models you created to show whether or not each suspect could have made hydrofluoric acid. Consider how the information presented in the evidence cards might relate to your models. Then, select the pieces of evidence you are going to use in your argument from the list below.
	 Evidence Card A: When the police searched Pat's house, they found some calcium sulfate (CaSO₄).
	O Evidence Card B: The police did not find hydrofluoric acid (HF), sulfuric acid (H ₂ SO ₄), or calcium fluoride (CaF ₂) in Pat's house.
	O Evidence Card C: When the police searched Tracy's house, they found some purified water (H ₂ O).
	O Evidence Card D: The police did not find hydrofluoric acid (HF), fluorine (F ₂), or any other unusual substances in Tracy's house.
	O Evidence Card E: According to Dr. Yung, calcium sulfate (CaSO ₄) is a substance commonly used in sculpting.
	\bigcirc Evidence Card F: According to Dr. Yung, oxygen (O_2) is a colorless, odorless gas that is commonly found in air.
	O Evidence Card G: The chemical supply company confirmed the delivery of sulfuric acid (H,SO ₂) and calcium fluoride (CaF ₂) to Pat's house.
	O Evidence Card H: The chemical supply company confirmed the delivery of purified water (H,O) and fluorine (F,) to Tracy's house.
	2,

Chemical Reactions-Lesson 4.3

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3.	Write a scientific argument that addresses the question: Which suspect is most likely to have made the hydrofluoric acid? First, state your claim. Then, use evidence to support your claim. For each piece of evidence you use, explain how the evidence supports your claim. As you write, refer back to your completed Modeling Tool sheets and Suspect Evidence Cards.						

Chemical Reactions—Lesson 4.3 © 2016 The Regents of the University of California Chemical Reactions—Lesson 4.3 © 2016 The Regents of the University of California

Science Seminar Reflection

How do students use literacy practices to:

- make and evaluate claims
- gather evidence
- justify arguments with relevant reasoning



Considering claims & evidence

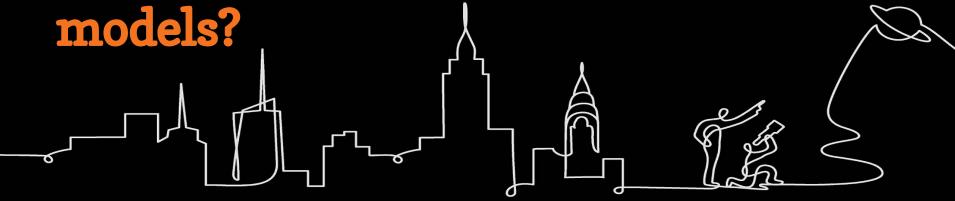


Participating in the Science Seminar



Writing an argument

Reflect-Type-Chat! Share and Learn
How can you adapt the science
seminar or how have you adapted
for remote and hybrid instructional



Science seminar and distance learning

- If possible, have an online discussion. Set discussion normals.
- If needed, use a collaborative document (Google docs, Padlet, Jamboard) for virtual, asynchronous student discussion
- Use existing student supports (sentence stems, science seminar guidelines)
- Create digital evidence cards (Google slides, Padlet, Jamboard) and have students sort them
- Have students share their screens to visually support the evidence they share
- Consider leading seminars in smaller groups
- See @Home Units for guidance
- Have students prepare independently or in partners



Plan for the day

- Framing the day
 - Welcome and introductions
- Shifts in Science Instruction
 - Revisiting the Amplify Science approach
 - Revisiting Resources
 - Science and Engineering practices
- Science and Literacy
 - Embedded literacy supports
 - Embedded literacy strategies
- Applying the Strategies through the Science Seminar
 - Evaluating claims, gathering evidence, and justifying arguments with reasoning
- Reflection and closing

Revisiting Our Objectives:

- Focus on the shifts in emphasizing the Science and Engineering practices which require students to read and write like scientists.
- Identify embedded strategies for explicit literacy support.
- Understand how these strategies help students make and evaluate claims, gather evidence, and justify arguments with relevant reasoning.
- Engage in reflection on how a science seminar structures opportunities for verbal and written argumentation.

Revisiting our objectives

Do you feel ready to...

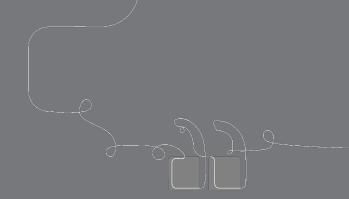
- Focus on the shifts in emphasizing the Science and Engineering practices which require students to read and write like scientists.
- Identify embedded strategies for explicit literacy support.
- Understand how these strategies help students make and evaluate claims, gather evidence, and justify arguments with relevant reasoning.
- Engage in reflection on how a science seminar structures opportunities for verbal and written argumentation.

1- I'm not sure how I'm going to do this!

3- I have some good ideas but still have some questions.

5- I have a solid plan for how to make this work!





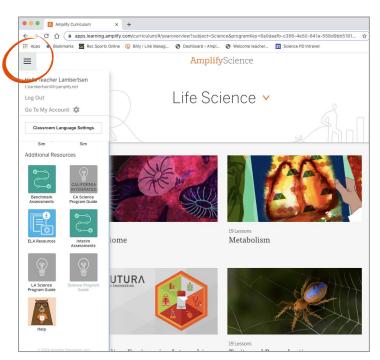
Questions?

Amplify Science Program Hub

A new hub for Amplify Science resources

- Videos and resources to continue getting ready to teach
- Amplify@Home resources
- Keep checking back for updates

science.amplify.com/programhub



New York City Resources Site

https://amplify.com/resources-page-for-nyc-6-8/



Amplify.

Amplify Science Resources for NYC (6-8)

Welcome! This site contains supporting resources designed for the New York City Department of Education Amplify Science adoption for grades 6–8.



Educator Spotlight Submission

Site Resources

- Login information
- Pacing guides
- Getting started guide
- NYC Companion Lessons
- Resources from PD sessions
- And much more!

Educator Spotlight Submission

20-21 Login Update

Professional learning opportunities

Calling all NYC DOE educators! Do you know an educator who has gone above and beyond? Would you like to highlight your teaching experience for others? Submit nominations here to see them featured as a spotlight in a future edition of our monthly newsletter and on our Instagram pages!

Introduction

started resources

Additional Amplify resources



Program Guide

Glean additional insight into the program's structure, intent, philosophies, supports, and flexibility.

https://my.amplify.com/programguide/content/national/welcome/science/

Amplify Help

Find lots of advice and answers from the Amplify team.

my.amplify.com/help

Additional Amplify Support

Customer Care

Seek information specific to enrollment and rosters, technical support, materials and kits, and teaching support, weekdays 7AM-7PM EST.



scihelp@amplify.com



800-823-1969



Amplify Chat

When contacting the customer care team:

- Identify yourself as an Amplify Science user.
- Note the unit you are teaching.
- Note the type of device you are using (Chromebook, iPad, Windows, laptop).
- Note the web browser you are using (Chrome or Safari).
- Include a screenshot of the problem, if possible.
- Copy your district or site IT contact on emails.